

Structurally Characterized Coinage-Metal–Ethylene Complexes

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A few data in Table 1 on page 512 of the original article^[1] are erroneous; the correct Table 1 is given below with the changes indicated in bold and italic.

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Table 1. Selected structural parameters and NMR spectroscopic data for coinage-metal–ethylene complexes: CN = coordination number at the coinage metal; for $d(\text{M}-\text{C})$ and $d(\text{C}=\text{C})$, average bond lengths are given if more than one is present; ^1H and ^{13}C NMR data in CDCl_3 unless otherwise noted.

Compound	CN	$d(\text{M}-\text{C})$ [Å]	$d(\text{C}=\text{C})$ [Å]	$\delta(^1\text{H})$ [ppm]	$\delta(^{13}\text{C})$ [ppm] ¹ J_{CH} [Hz]	Ref.
[HB{3,5-(CF ₃) ₂ Pz} ₃]Cu(C ₂ H ₄)	4	2.022(6)	1.325(9)	4.96	89.5/161 (C₆D₁₂)	[82]
[HB{3-(CF ₃)Pz} ₃]Cu(C ₂ H ₄)	4	2.009(9)	1.34(1)	4.80	85.8/158 (C₆D₆)	[82]
[HB{3-(CF ₃)-5-(Ph)Pz} ₃]Cu(C ₂ H ₄)	4	2.012(11)	1.30(1)	4.91	85.7/159 (C₆D₆)	[82]
[MeB{3-(CF ₃)Pz} ₃]Cu(C ₂ H ₄)	4	2.038(3)	1.334(4)	4.80	85.4/161	[103]
[Bu ₂ P(NSiMe ₃) ₂]Cu(C ₂ H ₄)	3	1.987(3)	1.362(6)	3.48 (C ₆ D ₆)	73/158	[104]
[HC{(Me)C(2,6-Me ₂ C ₆ H ₃ N) ₂ }Cu(C ₂ H ₄)	3	1.989(2)	1.365(3)	2.919 (C ₆ D ₆)	74.7/158	[107]
[N{(C ₃ F ₇)C(C ₆ F ₅)N ₂ }Cu(C ₂ H ₄)	3	2.014(3)	1.364(4)	3.86	86.1	[109]
[(tmen)Cu(C ₂ H ₄)]ClO ₄	3	1.96(1)	1.36(1)	4.16 (CD ₃ COCD ₃)		[114]
[(bpy)Cu(C ₂ H ₄)]ClO ₄	3	1.992(12)	1.353(15)	4.92 (CD ₃ COCD ₃)		[111]
[(phen)Cu(C ₂ H ₄)]ClO ₄	3	2.010(13)	1.361(22)	5.00 (CD ₃ COCD ₃)		[111]
[Cu(NH(py) ₂)(C ₂ H ₄)]ClO ₄	3	2.019 (3)	1.359 (7)	4.70 (CD ₃ COCD ₃)		[110]
[Cu(Hhq) ₂ (C ₂ H ₄)]ClO ₄	3	2.046(6)	1.32(1)	5.22 (CD ₃ COCD ₃)		[113]
{[Cu ₂ (bpm) ₂ (C ₂ H ₄)(Me ₂ CO)](ClO ₄) ₂ } _n	3	2.022(6)	1.30(1)			[112]
β-(C ₂ H ₄)CuAlCl ₄	4	2.089	1.34		105 (solid)	[86]
(C ₂ H ₄) ₂ CuAlCl ₄	5 ^[a]	2.137	1.34		109 (solid)	[86]
[Cu(C ₂ H ₄) ₃][Al{OC(CF ₃) ₃ } ₄]	3	2.147(1)	1.331(16)	5.47 (CD ₂ Cl ₂)	110.0	[87]
[HB{3,5-(CF ₃) ₂ Pz} ₃]Ag(C ₂ H ₄)	4	2.301(7)	1.298(14)	5.56 (C ₆ D ₁₂)	104.9/164	[83]
[MeB{3-(CF ₃)Pz} ₃]Ag(C ₂ H ₄)	4	2.2822(18)	1.340(4)	5.47	104	[125]
[MeB{3-(C ₂ F ₅)Pz} ₃]Ag(C ₂ H ₄)	4	2.301(2)	1.314(4)	5.48 (CD ₂ Cl ₂)	105.5/161	[124]
[MeB{3-(Mes)Pz} ₃]Ag(C ₂ H ₄)	4	2.27(3)	1.323(12)	3.42	95.4	[126]
[PhB{3-(CF ₃)Pz} ₃]Ag(C ₂ H ₄)	3	2.265(4)	1.296(7)	4.72	101.7	[125]
[HC{(CF ₃)C(3,5-(CF ₃) ₂ C ₆ H ₃ N) ₂ }Ag(C ₂ H ₄)	3	2.22(2)	1.392(19)	3.78 (C ₆ D ₆)		[127]
[{Ag(C ₂ H ₄) ₃][Al{OC(CF ₃) ₃ } ₂][Al{OC(CF ₃) ₃ } ₃] ₄]	3	2.289(8)	1.351(15)	5.54 (CD ₂ Cl ₂)	112.2	[128]
[Ag(C ₂ H ₄) ₃][Al{OC(CF ₃) ₃ } ₃] ₄]	3	2.396(5)	1.304(10)	5.77 (CD ₂ Cl ₂)	116.4	[85]
[HB{3,5-(CF ₃) ₂ Pz} ₃]Au(C ₂ H ₄)	3	2.102(6)	1.380(10)	3.81	63.7/165	[84]
[HB{3-(CF ₃)-5-(Ph)Pz} ₃]Au(C ₂ H ₄)	3	2.095(5)	1.387(9)	3.69	59.3/165	[84]
[Au(C ₂ H ₄) ₃][SbF ₆]	3	2.268(5)	1.364(7)	4.94 (CD ₂ Cl ₂)	92.7	[88]
C ₂ H ₄			1.313(1) (X-ray)	5.40 (CD ₂ Cl ₂)	123.3/156	[82,151,152]
			1.3369(16) (electr. diff.)			

[a] Three-coordinate Cu has two weak Cu···Cl contacts.

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